

## Ocr F325 June 2010 Question Paper

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### Ocr F325 June 2010 Question

F325 Mark Scheme June 2010 6 Question Expected Answers Marks Additional Guidance d HNO<sub>3</sub> + CH<sub>3</sub>CH<sub>2</sub>COOH ⇌ CH<sub>3</sub>CH<sub>2</sub>COOH<sub>2</sub><sup>+</sup> + NO<sub>3</sub><sup>-</sup> acid 1 base 2 acid 2 base 1 2 State symbols NOT required ALLOW 1 AND 2 labels the other way around. ALLOW 'just acid' and 'base' labels throughout if linked by lines so that

### Mark Scheme for June 2010 - AQA, OCR and Edexcel

† Write your answer to each question in the space provided. Additional paper may be used if

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necessary but you must clearly show your Candidate Number, Centre Number and question number(s). INFORMATION FOR CANDIDATES † The number of marks is given in brackets [ ] at the end of each question or part question.

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### **F325 June 2010**

F325 Mark Scheme June 2010 7 Question Expected Answers Marks Additional Guidance 4 a i  
Complete circuit (with voltmeter) and salt bridge linking two half-cells □ Pt electrode in solution of  
Fe<sup>2+</sup>/Fe<sup>3+</sup> □ Ag in solution of Ag<sup>+</sup> □ 3 DO NOT ALLOW 'solution of a silver halide', e.g. AgCl (as these  
are insoluble) but

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F325 Mark Scheme June 2011 1 Question Answer Mark Guidance 1 (a) (The enthalpy change that accompanies) the formation of one mole of a(n ionic) compound from its gaseous ions (under standard conditions) 2 IGNORE 'Energy needed' OR 'energy required' ALLOW as alternative for compound: lattice, crystal, substance, solid Note: 1st mark requires 1 mole

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OCR (Oxford Cambridge and RSA) is a leading UK awarding body, providing a wide range of qualifications to meet the needs of candidates of all ages and abilities. OCR qualifications ... F325 Mark Scheme June 2012 Question Answer Marks Guidance 2 (a) (iii) FIRST, CHECK THE ANSWER ON ANSWER LINE . IF.

### **Advanced GCE Unit F325: Equilibria, Energetics and Elements**

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### **Advanced GCE Unit F325: Equilibria, Energetics and Elements**

F325 Mark Scheme June 2010 8 Question Expected Answers Marks Additional Guidance c A fuel cell converts energy from reaction of a fuel with oxygen into a voltage/electrical energy □ 5 ANNOTATIONS MUST BE USED ALLOW combustion for reaction of fuel with oxygen/reactants ALLOW a fuel cell requires constant supply of fuel

### **3 ALLOW F 1450 B 736 G E 76 D -642 FIVE FOUR THREE b ALLOW ...**

F325 Mark Scheme June 2011 1 Question Answer Mark Guidance 1 (a) (The enthalpy change that accompanies) the formation of one mole of a(n ionic) compound from its gaseous ions (under standard conditions) 2 IGNORE 'Energy needed' OR 'energy required' ALLOW as alternative for compound: lattice, crystal, substance, solid Note: 1st mark requires 1 mole

### **Advanced GCE Unit F325: Equilibria, Energetics and Elements**

F325 Mark Scheme June 2014 3 Question Answer Marks Guidance 1 (a) (i)  $2K^+(g) + S^{2-}(g) + e^-$   
 $2K^+(g) + S^{2-}(g)$   $2K(g) + S(g)$  3 Mark each marking point independently Correct species AND state symbols required for each mark For  $S^{2-}$ , DO NOT ALLOW  $S^{-2}$  For  $e^-$ , ALLOW  $e$  For  $e^-$  only, IGNORE

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any state symbols added ALLOW k and s It can be very difficult distinguishing K from k; S from s

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