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**Building Blocks of
Matter**

Particle Mass number
19. Proton 20. Neutron
21. Electron 10

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law of conservation of mass. states that mass is neither created nor destroyed during ordinary reactions or physical changes.

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of an atom in a chemical compound to attract electrons from another atom in the compound is called _____. 17. The energy required to remove one electron from an atom is called its _____. 18.

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Answers

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are composed of the
same two elements,
then the ratio of the
masses of the second
element combined with
a certain mass of the
first element is always
a ratio of small whole

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numbers; textbook: if two or more different compounds are composed of the same two elements with a certain mass of the first element is ...

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3:3 mol H₂; or their reciprocals 7. Given the following equation:
 $4\text{NH}_3(\text{g}) + 6\text{NO}(\text{g}) \rightarrow 5\text{N}_2(\text{g}) + 6\text{H}_2\text{O}(\text{g})$
1 mol NO:1 mol H₂O a. What

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is the mole ratio of NO to H₂O? 3 mol NO:2 mol NH₃ b. What is the mole ratio of NO to NH₃? 0.360 mol c. If 0.240 mol of NH₃ react according to the above equation, how many moles of NO will be consumed? 8.

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Township Public
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REVIEW The Periodic
Law SECTION 1 SHORT
ANSWER Answer the
following questions in
the space provided. 1.
c In the modern
periodic table,
elements are ordered
(a) according to
decreasing atomic
mass.

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